

JARAMOGI OGINGA ODINGA UNIVERSITY OF SCIENCE AND TECHNOLOGY

SCHOOL OF ENGINEERING AND TECHNOLOGY

UNIVERSITY EXAMIMATION FOR THE DEGREE IN SCIENCE IN RENEWABLE ENERGY TECHNOLOGY AND MANAGEMENT

2ND YEAR 1ST SEMESTER 2023/2024 ACADEMIC YEAR

CENTRE: MAIN CAMPUS

COURSE CODE: TEB 1201

COURSE TITLE: MATERIAL SCIENCE I

EXAM VENUE:

STREAM: BSc. REN ENGY TEC & MGT

DATE: /12/2023 EXAM SESSION:

DURATION: 2 HOURS

Instructions

- 1. Answer question 1 (Compulsory) and ANY other two questions
- 2. Candidates are advised not to write on question paper
- 3. Candidates must hand in their answer booklets to the invigilator while in the examination room.

Question One (Compulsory) (30 Marks)

- a) Explain **WHY** materials bonded in with covalent bonds have limited ductility while those bonded with metallic bonds show good ductility (**2 Marks**)
- b) Explain the difference between the terms materials science and materials engineering. (3 Marks)
- c) Using a well labeled diagram illustrate the principles of atomic bonding showing the presence of attractive and repulsive forces (4 Marks)
- d) Define the following terms (5 Marks)
 - 1. Binding energy
 - 2. Electronegativity
 - 3. Crystalline materials
 - 4. Composition

e) Out line **FOUR** considerations that a designer has to consider for satisfactory fatigue life during design procedure (**4 Marks**)

- f) Discuss two materials of the future that would have a significant influence on many of our recent technologies in production (4 Marks).
- e) A piece of copper originally 305 mm long is pulled in tension with a stress of 276 MPa. If the deformation is entirely elastic, find the resultant elongation? (3 Marks)
- g) Outline four information that can be obtained from the phase diagrams (4 Marks)

h) Enumerate three methods are widely used for the control and preventation of corrosion in metals (3 Marks).

QUESTION TWO (20 Marks)

- a) Discuss FOUR blocks that are used to describe the modern periodic table. (10 Marks)
- b) As an expert in material science recommend **FIVE** classifications of material structure an engineer would consider while characterizing materials. (**10 Marks**)

QUESTION THREE (20 Marks)

a) Apart from the mechanical properties of materials, outline other five properties that are important for an engineer, to enable him in selecting suitable metals for various jobs (5 Marks).

- b) Discus the following bonds (5 Marks)
 - 1. Ionic bond
 - 2. Covalent bond
- c) Draw and discuss the cooling curves for a binary system forming eutective solution. (6 Marks)
- d) What is heat treatment in material science engineering? (2 Marks)
- e) What is the use of time-temperature transformation (T-T-T) curves? (2 Marks)

QUESTION FOUR (20 Marks)

- a) Discuss five types of corrosion that occur on metal surfaces (10 Marks)
- b) Use a well labelled diagram discuss the entire range representing the iron-carbon phase diagram (10 Marks)

QUESTION FIVE (20 Marks)

- a) Using a well labelled diagram sketch a stress-strain curve obtained from tensile test. Discuss any five properties obtained from the curve (**10 Marks**)
- b) Discuss four non-destructive testing methods that are used in material science. (10 Marks)