

JARAMOGI OGINGA ODINGA UNIVERSITY OF SCIENCE AND TECHNOLOGY SCHOOL ENGINEERING SCIENCES, HEALTH SCIENCES, SPATIAL PLANNING, BIOLOGICAL SCIENCES, HORTLICULTURAL STUDIES FIRST YEAR SECOND SEMESTER EXAMINATIONS FOR THE DEGREE OF BACHELOR OF SCIENCE EDUCATION WITH IT

SCH 3112/SPB 9103: BASIC ORGANIC CHEMISTRY **UNIVERSITY EXAMINATIONS: 2021/2022 ACADEMIC YEAR**

ANSWER ALL QUESTIONS IN SECTION A AND ANY TWO QUESTIONS FROM **SECTION B**

SECTION A: ANSWER ALL QUESTIONS

QUESTION 1

a) Give the IUPAC names of the following compounds;

(10 marks)

(i)
$$CH_3(CH_2)_8CH_3$$
 (ii) $CH_2=CH(CH_2)_2CH=CH=CCH=CH_2$

Br

(x)
$$CH_3$$
- $C(I)_2$ - $C(I)_2$ - CH_2 - CH_2 - CH_3

b) Draw the structures of the following compounds;

(10 marks)

- Prop-2-enol i)
- ii) 3,4-dibromo-3,4-dichlorohept-1-yne
- iii) Nonane
- iv) Octanol
- v) Cycohexane

- c) Using TWO examples, give the classification of alkylhalides.
- (6 marks)
- d) Esterification is a reaction typical of alkanols and alkanoic acids. Illustrate this reaction using ethanol and propanoic acid. (4 marks)

SECTION B (40 MMARKS):

ANSWER ANY TWO QUESTIONS FROM THIS SECTION-**EACH QUESTION CARRIES 20 MARKS**

QUESTION 2

a) Give a brief explanation to the following observations:

(15 marks)

- (i) Boiling points of alkanes increase down along the homologous series.
- (ii) Branched alkylhalides have lower boiling points compared to the corresponding straight-line derivatives.
- (iii) Lower alcohols are miscible with water.
- (iv) C-H-C bond angle is 109.5° in alkanes
- (v) Alkenes decolourise bromine water.
- b) Define the following terms;

(5 marks)

- (i) Catenation
- (ii) Homologous series
- (iii) Functional group
- (iv) Cracking
- (v) Zwitter ion

QUESTION 3

Give the products of the following reactions; a)

(12 marks)

(ii)
$$CH_4 + Cl_2 \xrightarrow{hv}$$

(iv) H-C
$$\equiv$$
 C-Li $^+$ + CH₃-CH₂-Cl

(vi)
$$CH_2 = CH_2 + HC1$$

b) Carbon is said to be a very unique element. Discuss this statement.

QUESTION 4

a) Discuss the structure and bonding in methane

(6 marks)

(8 marks)

b) Give the mechanism for the following reactions and name the major products; (6 marks)

- (i) CH_3 -CH=CH- CH_3 + Br_2 $\xrightarrow{H_2O}$
- CH_3 - $CH_3 + I_2$ OH (ii)
- Complete the following reactions; (c)

(8 marks)

(ii)
$$CH_3CH=CH_2 + O_3$$
 Zn, H_2O

(ii)
$$CH_3CH=CH_2 + O_3$$
 OH^-/H_2O
(iii) $CH_3 - C = O$

(iv)
$$CH_3 - C \equiv C - H + H_2O$$
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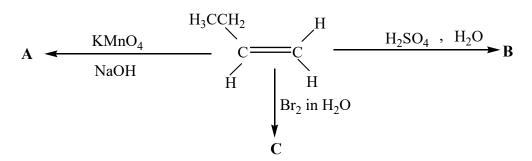
QUESTION 5

a) Give the mechanism for the reactions below; (10 marks)

(i)
$$CH_3CH=CH_2 + O_3$$
 Zn, H_2O

(ii)
$$CH_3 - C \equiv C - H + H_2O$$
..

- Briefly discuss the significance of studying Organic Chemistry. (5 marks) b)
- c) Give products (A-C) to complete the following scheme: (3 marks)



Explain any **FOUR** uses of alkylhalides. d)

(2 marks)

 $E^{**********}N^{*******}D$