

JARAMOGI OGINGA ODINGA UNIVERSITY OF SCIENCE AND TECHNOLOGY SCHOOL OF BIOLOGICAL AND PHYSICAL SCIENCES UNIVERSITY EXAMINATION FOR THE DEGREE OF BACHELOR OF EDUCATION (SCIENCE)

1ST YEAR 2ND SEMESTER 2017/18

MAIN REGULAR

COURSE CODE: SCH 103

COURSE TITLE: Basic Organic Chemistry

EXAM VENUE: STREAM: (BED SCI)

DATE: EXAM SESSION:

TIME: 2:00HRS

Instructions:

- 1. Answer question 1 (Compulsory) in Section A and ANY other 2 questions in Section B.
- 2. Candidates are advised not to write on the question paper.
- 3. Candidates must hand in their answer booklets to the invigilator while in the examination room

INSTRUCTIONS: Answer Question 1 and any other TWO questions

QUESTION ONE (Compulsory) (30 marks)

1. (a) Define the following terms:

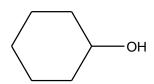
- (i) Isomers
- (ii) Aromatic hydrocarbon
- (iii) Homologous series
- (iv)Functional group
- (v) Markovnikovs rule

[10 marks]

- b) Draw structural formulas corresponding to the following names:
 - (i) 5-bromo-3-ethylhexanoic acid
 - (ii) 3-methyl-2-hexen-4-yne
 - (iii) 2,3,4-trimethyl-4-propylheptane
 - (iv) 2,5-dimethyl-1,5-hexadiene

[8 marks]

(c) Give the name of each of the following organic compound:



[10 marks]

(d) Complete the following reactions by giving the main organic product formed:

(i)

OH
$$\frac{\text{conc H}_2\text{SO}_4}{180^{\circ}\text{C}}$$

(ii)

[4 marks]

QUESTION TWO (20 marks)

- 2. (a) Give the type of hybridization present in the following aliphatic hydrocarbons:
 - (i) Pent-2-ene
 - (ii) Ethane
 - (iii) Propyne

[3 marks]

- (b) Differentiate between Tollen's and Benedict's test for aldehydes. [6 marks]
- (c) Using examples, discuss any **FOUR** reactions involving alkenes. [8 marks]
- (d) Propose structures for the following;
 - i) A two carbon ester
 - ii) A three carbon amide
 - iii) A four carbon ether.

[3 marks]

QUESTION THREE (20 marks)

3. (a) State any **TWO** physical properties of alkanes.

[2 marks]

(b) Consider the following structure

i) How many primary carbon atoms does it have? [1 mark]
 ii) How many secondary carbon atoms does it have [1 mark]
 iii) How many tertiary carbon atoms does it have? [1 mark]
 iv) Write down the molecular formula. [2 marks]
 v) Convert the skeletal structure into condensed structure [2 marks]

(c) State the uniqueness of carbon in organic chemistry.

[4 marks]

(e) Draw the structure of butyl benzoate.

[1 marks]

(f) Complete the following reactions by giving the main organic product formed:

(i)

2-methyl-2-propanol + conc.
$$H_2SO_4$$
/excess heat /180 °C \rightarrow

[2 marks]

(ii)

(iii) 2-chloropropanoic acid + CaCO₃ →

[2 marks]

QUESTION FOUR (20 marks)

4. (a) State **TWO** uses of each of the following organic compounds;

- (i) Carboxylic acids
- (ii) Alcohols
- (iii) Esters

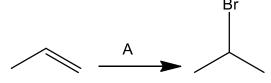
(iv) Alkenes [8 marks]

- (b) Write down the products formed when 2-methylpropanol reacts with;
 - (i) Excess Conc. H₂SO₄ / heat / 180 °C
 - (ii) $K_2Cr_2O_7 / H+/ \Delta$
 - (iii) SOCl₂
 - (iv) Na [8 marks]
- (c) Give the product of the reaction of ethanal with;
- (i) LiAlH₄
- (ii) Fehling's reagent [4 marks]

QUESTION FIVE (20 marks)

- 5. (a) The following name is incorrect. Draw the molecule and give its correct name.

 1-methyl-2-cyclopentene [4 marks]
 - (b) Fill in the missing reagent needed for the following reactions to take place ad name the product



[3 marks]

- (c) Complete the following organic reactions by giving the main organic product(s):
 - (i) 3-methylpentene + HBr \rightarrow [2 marks]
 - (ii) 3-bromobenzoic acid + 2-propanol \rightarrow [2 marks]

(d) Briefly discuss the reactions of alkyl halides. [6 marks]

(e) State any **THREE** features of a homologous series [3 marks]