JARAMOGI OGINGA ODINGA UNIVERSITY OF SCIENCE AND TECHNOLOGY SCHOOL OF BIOLOGICAL AND PHYSICAL SCIENCES

UNIVERSITY EXAMINATION FOR THE DEGREE OF BACHELOR OF EDUCATION (SCIENCE)
$1^{\text {ST }}$ YEAR $^{\text {ND }}{ }^{\text {ND }}$ SEMESTER 2017/18
MAIN REGULAR

COURSE CODE: SCH 103
COURSE TITLE: Basic Organic Chemistry
EXAM VENUE:
STREAM: (BED SCI)
DATE: EXAM SESSION:

TIME: 2:00HRS

## Instructions:

1. Answer question 1 (Compulsory) in Section $A$ and ANY other 2 questions in Section B.
2. Candidates are advised not to write on the question paper.
3. Candidates must hand in their answer booklets to the invigilator while in the examination room

# INSTRUCTIONS: Answer Question 1 and any other TWO questions QUESTION ONE (Compulsory) (30 marks) 

1. (a) Define the following terms:
(i) Isomers
(ii) Aromatic hydrocarbon
(iii) Homologous series
(iv)Functional group
(v) Markovnikovs rule
[10 marks]
b) Draw structural formulas corresponding to the following names:
(i) 5-bromo-3-ethylhexanoic acid
(ii) 3-methyl-2-hexen-4-yne
(iii) 2,3,4-trimethyl-4-propylheptane
(iv) 2,5-dimethyl-1,5-hexadiene
[8 marks]
(c) Give the name of each of the following organic compound:
(i)

(ii)

(iii)

(iv)

(v)

[10 marks]
(d) Complete the following reactions by giving the main organic product formed:
(i)

(ii)


## QUESTION TWO (20 marks)

2. (a) Give the type of hybridization present in the following aliphatic hydrocarbons:
(i) Pent-2-ene
(ii) Ethane
(iii) Propyne
[3 marks]
(b) Differentiate between Tollen's and Benedict's test for aldehydes.
(c) Using examples, discuss any FOUR reactions involving alkenes.
(d) Propose structures for the following;
i) A two carbon ester
ii) A three carbon amide
iii) A four carbon ether.

## QUESTION THREE (20 marks)

3. (a) State any TWO physical properties of alkanes.
(b) Consider the following structure

i) How many primary carbon atoms does it have?
ii) How many secondary carbon atoms does it have
iii) How many tertiary carbon atoms does it have?
iv) Write down the molecular formula.
v) Convert the skeletal structure into condensed structure
(c) State the uniqueness of carbon in organic chemistry.
(e) Draw the structure of butyl benzoate.
(f) Complete the following reactions by giving the main organic product formed:
(i)

2-methyl-2-propanol + conc. $\mathrm{H}_{2} \mathrm{SO}_{4} /$ excess heat $/ 180^{\circ} \mathrm{C} \rightarrow$
[2 marks]
(ii)

(iii) 2-chloropropanoic acid $+\mathrm{CaCO}_{3} \rightarrow$

## QUESTION FOUR (20 marks)

4. (a) State TWO uses of each of the following organic compounds;
(i) Carboxylic acids
(ii) Alcohols
(iii) Esters
(iv) Alkenes
(b) Write down the products formed when 2-methylpropanol reacts with;
(i) Excess Conc. $\mathrm{H}_{2} \mathrm{SO}_{4} /$ heat / $180{ }^{\circ} \mathrm{C}$
(ii) $\quad \mathrm{K}_{2} \mathrm{Cr}_{2} \mathrm{O}_{7} / \mathrm{H}+/ \Delta$
(iii) $\mathrm{SOCl}_{2}$
(iv) Na
(c) Give the product of the reaction of ethanal with;
(i) $\mathrm{LiAlH}_{4}$
(ii) Fehling's reagent

## QUESTION FIVE (20 marks)

5. (a) The following name is incorrect. Draw the molecule and give its correct name.

1-methyl-2-cyclopentene
[4 marks]
(b) Fill in the missing reagent needed for the following reactions to take place ad name the product

(c) Complete the following organic reactions by giving the main organic product(s):
(i) 3-methylpentene $+\mathrm{HBr} \rightarrow$
[2 marks]
(ii) 3-bromobenzoic acid +2 -propanol $\rightarrow$
[2 marks]
(d) Briefly discuss the reactions of alkyl halides.
(e) State any THREE features of a homologous series

